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# Tannins and Related Compounds from *Erodium*moschatum (L.) L'Her

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#### **ABSTRACT**

Erodium moschatum is a newly naturalized plant in Taiwan. From the aqueous acetone extract of the fresh herb, seventeen tannins and related compounds were isolated. They included five phenolcarboxylic acids and ester including: protocatechuic acid (1), gallic acid (2), methyl gallate (3), caffeic acid (4), brevifolincarboxylic acid (5); four gallotannins: 3-O-galloylshikimic acid (6), 3,4-di-O-galloylshikimic acid (7), 3,5-di-O-galloylshikimic acid (8), 1-O-galloyl-β-D-glucose (9); six ellagitannins and other related compounds which include corilagin (10), furosin (11), geraniin (12), acetonylgeraniin A (13), methyl gallate 3-O-β-D-glucoside (14), gallic acid 3-O-β-D-(6'-O-galloyl)-glucoside (15) and two flavonoids: kaempferol (16), quercetin (17). These structures were identified on the basis of their physical data and spectroscopic evidence.

Key words: Erodium moschatum, Geraniaceae, hydrolyzable tannin, flavonoid.

#### INTRODUCTION

Since ancient times, Geraniaceous plants including genus *Geranium* and *Erodium* were used for the treatment of diarrhea<sup>(1,2)</sup>. The active principles are tannins and the main component has been shown to be geraniin<sup>(3)</sup>. In Taiwan, only the genus *Geranium* has been described in the "Flora of Taiwan"<sup>(4)</sup>. In 1994, Ou and Kao reported *Erodium moschatum* as a newly recorded plant in Taiwan<sup>(5)</sup>. *E. moschatum* was originally distributed in southern and western Europe, Africa, America, Indonesia and Japan, and naturalized at the Ching-Ch'ing farm, Nantou, Taiwan<sup>(5)</sup>. From

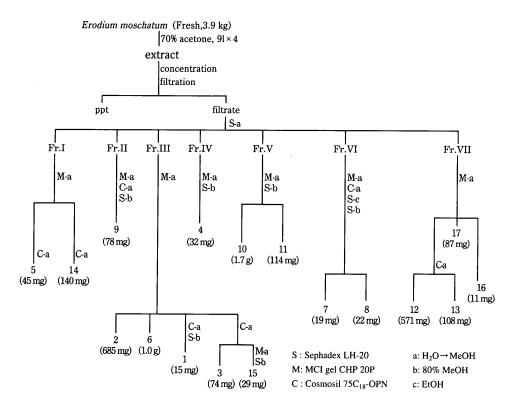
this plant, seventeen compounds were isolated. This paper describes the isolation procedures and structural elucidation of these compounds.

#### MATERIALS AND METHODS

#### I. Instruments and Reagents

Melting points were determined with a Fisher-Johns melting-point apparatus and were uncorrected. Optical rotation was measured with a Jasco DIP-140 polarimeter. Mass spectra were obtained on a JEOL JMS HX110 spectrometer. 

<sup>1</sup>H(300 MHz) and <sup>13</sup>C(75 MHz) NMR spectra



**Scheme 1.** Isolation of tannins and related compounds from *E. moschatum* 

were recorded on a Bruker AM-300WB spectrometer, using the solvent peak as reference standard. Column chromatography was carried out with Sephadex LH-20(25-150  $\mu$ m, Pharmacia Fine Chemical Co. Ltd.), MCI gel CHP 20P (75-150  $\mu$ m, Mitsubishi Chemical Industries, Ltd.), and Cosmosil 75 C<sub>18</sub>-OPN columns. Thin-layer chromatography was conducted on precoated Kiesel gel 60 F<sub>254</sub> plates (0.2 mm, Merck) and Cellulose F plates (0.1 mm, Merck) and spots were detected by UV illumination before spraying with ethanolic ferric chloride.

#### II. Plant Material

The fresh Erodium moschatum were collected at the Ching-Ch'ing Farm, Nantou, Taiwan, Republic of China, and verified by Mr. Muh-Tsuen Kao, Department of Botany, National Taiwan University. The voucher specimen is deposited in the National Laboratories of Foods and Drugs, Department of Health, Executive Yuan, Republic of China.

#### III. Extraction and Separation

The fresh *Erodium moschatum* (3.9 kg), as shown in Scheme 1, were extracted four times with 70% acetone (36 L) at room temperature. The acetone was removed by evaporation under reduced pressure (ca 40°C), and the resulting precipitate removed by filtration. The filtrate, after concentration, was subjected to a Sephadex LH-20 column chromatography using H<sub>2</sub>O containing increasing amounts of MeOH (0-100%) to yield seven fractions (Fr.I-VII). Repeated column chromatography of each fraction over Sephadex LH-20 (80% MeOH), MCI gel CHP 20P (0-100% MeOH) and Cosmosil 75 C<sub>18</sub>-OPN (0-100% MeOH) yielded the compounds 1-17.

#### Protocatechuic Acid (1)

White needles (H<sub>2</sub>O), mp 197-199°C, FeCl<sub>3</sub> reagent: dark green. IR (KBr)  $\upsilon_{max}$  cm<sup>-1</sup>: 2200-3500 (OH), 1650 (C = O). <sup>1</sup>H-NMR (acetone-d<sub>6</sub>):  $\delta$  6.90 (1H, d, J = 8.0 Hz, H-5), 7.47 (1H,dd, J = 2.0, 8.0 Hz, H-6), 7.53 (1H, d, J = 2.0 Hz, H-2).

Caffeic Acid (4)

$$R_1$$

$R_1$	$R_2$
1. COOH	H
2. COOH	OH
3. COOCH <sub>3</sub>	OH
4 CH=CH-COOH	н

5.

$$R_3O$$
  $OR_2$ 

 $\begin{array}{lll} \text{6. } R_1 \text{=} G & R_2 \text{=} R_3 \text{=} H \\ \text{7. } R_1 \text{=} R_2 \text{=} G & R_3 \text{=} H \\ \text{8. } R_1 \text{=} R_2 \text{=} G & R_2 \text{=} H \end{array}$ 

#### Acetonyl DHHDP

$$O = C \qquad C = O$$

$$O = HO \qquad OH$$

$$C = O \qquad OH$$

$$C = O \qquad CH_3$$

$$\begin{array}{c} R_4OCH_2 \\ OOR_2 \\ OOR_1 \end{array} \\ OOOD \\ O$$

 $R_1$  $R_3$  $R_2$  $R_4$ 9. Η Н Н Η Н 10. Η (R)HHDP **DHHDP** 11. Η Η 12. **DHHDP** (R)HHDP (R)HHDP 13. Acetonyl **DHHDP** 

G. HO O II C C C

HHDP

$$O=C$$
  $C=O$   $OH$ 

DHHDP O=C C=O

HO
OH
OH
OH
OH
OH
OH
OH
OH
OH

Structures of compounds 1-13

14. R<sub>1</sub>=CH<sub>3</sub>, R<sub>2</sub>=H 15. R<sub>1</sub>=H, R<sub>2</sub>=G

16. R = H17. R = OH

#### Structures of compounds 14-17

Colorless needles ( $H_2O$ ), mp 214-215°C. <sup>1</sup>H-NMR (acetone- $d_6 + D_2O$ ):  $\delta$  6.22 (1H, d,J = 16.2 Hz, H- $\alpha$ ), 6.83 (1H, d,J = 8.4 Hz, H-5), 6.96 (1H,dd,J = 2.1, 8.4 Hz,H-6), 7.13 (1H, d, J = 2.1 Hz, H-2), 7.49 (1H d, J = 16.2 Hz, H- $\beta$ ).

#### Brevifolincarboxylic Acid (5)

Yellow powder ( $H_2O$ -MeOH), mp>300°C, [ $\alpha$ ]  $^{25}$ -3.0° (c = 0.4,  $H_2O$ : acetone = 2:3).  $^1$ H-NMR (acetone- $d_6$  +  $D_2O$ ):  $\delta$  2.58 (1H, dd, J = 18.9, 1.8 Hz, H-5), 3.09 (1H, dd, J = 7.8, 18.9 Hz, H-5), 4.56 (1H, dd, J = 7.8 , 1.8 Hz, H-4), 7.45 (1H, s, H-3').

#### 3-O-Galloylshikimic Acid (6)

Colorless needles (H<sub>2</sub>O), mp 255°C (dec.), [ $\alpha$ ]  $_{D}^{25}$ -110.0° (c = 0.6, acetone).  $^{1}$ H-NMR (acetone-d<sub>6</sub>):  $\delta$  2.35 (1H, dd, J = 6.0, 18.4 Hz, H-2),2.78(1H, dd, J = 6.0, 18.4 Hz, H-2),3.98 (1H, dd, J = 4.2,7.0 Hz, H-4),4.56(1H, t, J = 4.2 Hz, H-4).

5),5.30(1H,m, H-3),6.82(1H, d, J = 4.2 Hz, H-6),7.12(2H, s, galloyl H).

#### 3,4-Di-O-galloylshikimic Acid (7)

Brown needles (H<sub>2</sub>O), mp 268-270°C, [ $\alpha$ ]  $^{25}_{D}$  168.3° (c = 1.0, acetone).  $^{1}$ H-NMR (acetone-d<sub>6</sub>):  $\delta$  2.49 (1H, dd, J = 5.4, 18.3 Hz, H-2),2.91(1H, dd, J = 5.9, 18.3 Hz, H-2), 5.34(1H, dd, J = 3.9, 8.0 Hz, H-4),5.56(1H, m, H-3),4.74(1H, t, J = 3.9 Hz, H-5), 6.92(1H, d, J = 3.9 Hz, H-6),7.04,7.06 (each 2H, s, galloyl H).

#### 3,5-Di-O-galloylshikimic Acid (8)

Brown needles (H<sub>2</sub>O), mp 237-238°C, [ $\alpha$ ]  $^{25}_{D}$  168.0°(c = 1.2, MeOH).  $^{1}$ H-NMR (acetone-d<sub>6</sub>):  $\delta$  2.49 (1H, dd, J = 4.8, 18.6 Hz, H-2), 3.04(1H, dd, J = 4.8, 18.6 Hz, H-2), 4.32(1H, dd, J = 4.2, 7.1 Hz, H-4),5.41(1H, m, H-3),5.79(1H, t, J = 4.2 Hz, H-5), 6.85(1H, d, J = 4.2 Hz, H-6),7.11,7.16 (each 2H, s, galloyl H).

#### Corilagin (10)

White amorphous powder ( $H_2O$ ), [ $\alpha$ ]  $^{25}_D$  190.2° (c = 0.8, acetone).  $^1H$ -NMR (acetone- $d_6$  +  $D_2O$ ):  $\delta$  4.05(1H, br s, glc H-2), 4.10 (1H, dd, J = 10.7,8.0 Hz, glc H-6), 4.43 (1H, br s, glc H-4), 4.49 (1H, dd, J = 10.7, 8.0 Hz, glc H-5), 4.79 (1H, br s, glc H-3), 4.84 (1H, t, J = 10.7 Hz, glc H-6), 6.38 (1H, br s, glc H-1), 6.67, 6.80 (each 1H, s, HHDP H), 7.08 (2H, s, galloyl H).

#### Furosin (11)

Yellow powder (H<sub>2</sub>O), mp 197-198°C (dec.),  $[\alpha]_{D}^{25}$ -142.1° (c=1.0, MeOH). <sup>1</sup>H-NMR (acetone-d<sub>6</sub>+D<sub>2</sub>O):  $\delta$  5.30 (1H, s, DHHDP H-1), 6.41 (1H, d, J=1.6 Hz, glc H-1), 6.48 (1H, s, DHHDP H-3), 7.18 (2H, s, galloyl H), 7.24 (1H, s, DHHDP H-3').

#### Geraniin (12)

Yellow powder ( $H_2O$ ), mp 218-221°C (dec.), [ $\alpha$ ]  $^{25}$ -147.8° (c=0.9, MeOH).  $^{1}$ H- $_{NMR}$  (acetone-d<sub>6</sub> + D<sub>2</sub>O):  $\delta$  4.28-4.54(1H, m, glc H-6),4.68-4.90 (2H in total , glc H-5,6), 5.15 (1H, s, DHHDP H-1), 5.40-5.60 (3H in total, glc H-2,3,4), 6.48 (1H, s, DHHDP H-3), 6.55 (1H, s, glc H-1), 6.63, 7.06

(each 1H, s, HHDP H), 7.14 (2H, s, galloyl H), 7.18 (1H, s, DHHDP H-3').

Acetonylgeraniin A (13)

White amorphous powder ( $H_2O$ ), mp 235-238 °C (dec.), [ $\alpha$ ]  $^{25}$  -91.6° (c=1.0, MeOH). Negative FABMS m/z: 991 [M-H]<sup>-</sup>. <sup>1</sup>H-NMR (acetone-d<sub>6</sub>):  $\delta$  2.17 (3H, s, -CH<sub>3</sub>), 2.97 and 3.46 (each 1H, d, J=15.6 Hz, -CH<sub>2</sub>-), 4.39 (1H, dd, J=6.0, 12.0 Hz, glc H-6), 4.74-4.86 (2H, m glc H-5, 6), 4.90 [1H, d, J=1.3 Hz, acetonyldehydrohexahydroxy-diphenyl (ADHHDP) H-1'], 5.42 (1H, m, glc H-3), 5.50 (1H, m, glc H-4), 5.54 (1H, m, glc H-2), 6.29 (1H, d, J=1.3 Hz, ADHHDP H-3'), 6.55 (1H, br s, glc H-1), 6.65, 7.06 (each 1H, s, HHDP H), 7.15 (2H, s, galloyl H), 7.21 (1H, s, ADHHDP H-3').

#### Methyl Gallate 3-O- $\beta$ -D-Glucoside (14)

White amorphous powder,  $[\alpha]_{D}^{25}$  -59.2° (c=0.5, 50% acetone), <sup>1</sup>H-NMR (acetone-d<sub>6</sub> + D<sub>2</sub>O):  $\delta$  3.75 (1H, dd, J=12.5,4.6 Hz, glc H-6), 3.77 (3H, s, -OCH<sub>3</sub>), 3.84 (1H, dd, J=1 2.5,4.6 Hz, glc H-6), 4.81 (1H, d, J=7.5 Hz, glc H-1), 7.22, 7.37 (each 1H, d, J=1.8 Hz, H-2,6). <sup>13</sup>C-NMR (acetone-d<sub>6</sub>):  $\delta$  52.3 (-OCH<sub>3</sub>), 61.5 (glc C-6), 70.3 (glc C-4), 73.0 (glc C-2), 76.5 (glc C-3), 77.3 (glc C-5), 103.2 (glc C-1), 111.0,112.5 (galloyl C-2,6), 121.0 (galloyl C-1), 140.9 (galloyl C-4), 146.1, 146.2 (galloyl C-3,5), 167.7 (-COO-).

Gallic Acid 3-O- $\beta$ -D-(6'-O-galloyl)-Glucoside (15)

White amorphous powder ( $\rm H_2O$ ), mp 175-178 °C, [ $\alpha$ ]  $^{25}_{D}$  -21.0° (c=1.0, acetone). Negative FABMS m/z: 483 [M-H]<sup>-</sup>. <sup>1</sup>H-NMR (acetone-d<sub>6</sub> + D<sub>2</sub>O):  $\delta$  3.4-3.6 (3H, m, glc H-2,3,4) ,3.9 (1H, m, glc H-5), 4.18 (1H, dd, J=7.2, 12.3 Hz, glc H-6), 4.68 (1H, dd, J=1.51,12.3 Hz, glc H-6), 4.96 (1H, d,J=7.4 Hz, glc H-1), 7.20 (2H, s, galloyl H), 7.26,7.45 (each 1H, d, J=1.8 Hz, galloyl H-2', 6'). <sup>13</sup>C-NMR (acetone -d<sub>6</sub> + D<sub>2</sub>O):  $\delta$  64.9 (glc C-6), 70.7 (glc C-4), 73.9 (glc C-2), 75.2 (glc C-5), 76.3 (glc C-3), 103.0 (glc C-1), 109.8 (2C, galloyl C-2,6), 110.4, 113.0 (galloyl C-2',6'), 120.7 (galloyl C-1), 121.9 (galloyl C-1'), 138.9(galloyl C-4), 140.5 (galloyl C-4'), 145.7(2C, galloyl C-3,5),

146.0, 146.4 (galloyl C-3',5'), 167.3 (-COO-), 169.8 (-COOH).

#### RESULTS AND DISCUSSION

The aqueous acetone extract of fresh E. moschatum was subjected to a combination of Sephadex LH-20, MCI gel CHP 20P and Cosmosil 75C<sub>18</sub>-OPN chromatographies with various solvent systems as shown in Scheme 1 to yield compounds 1-17. Compounds 1-5 were phenolcarboxylic acids or ester and were identified as protocatechuic acid (1)<sup>(6)</sup>, gallic acid (2), methly gallate (3), caffeic acid (4)<sup>(7)</sup>, brevifolincarboxylic acid (5)<sup>(8)</sup> by direct comparison of their physical and spectral profiles with literature values. Compounds 6, 7 and 8 gave a dark blue coloration with ferric chloride which is a characteristic of gallotannins. <sup>1</sup>H-NMR spectra of 6,7 and 8 showed signals in the aromatic region [6: $\delta$  7.12 (2H, s, galloyl H); 7: δ 7.04,7.06 (each 2H, s, golloyl H); 8:  $\delta$  7.11, 7.16 (each 2H, s, galloyl H)] indicated the presence of galloyl groups. These three compounds showed two double doublet signals near  $\delta$  2.5 and 3.0 (each 1H, J ca. 6, 18 Hz) indicating two H-2 of shikimic acid. Signals in the aliphatic region [6:  $\delta$  3.98 (1H, dd, J=4.2, 7.0 Hz, H-4), 4.56 (1H, t, J=4.2 Hz, H-5), 5.30 (1H, m, H-3), 6.82 (1H, d, J=4.2 Hz, H-6); 7: δ 4.74 (1H, t, J=3.9 Hz, H-5), 5.34 (1H, dd, J=3.9, 8.0 Hz, H-4), 5.56 (1H, m, H-3), 6.92 (1H, d, J=3.9 Hz, H-6); 8:  $\delta$  4.32 (1H, dd, J=4.2,7.1 Hz, H-4), 5.41 (1H, m, H-3), 5.79 (1H, t, J=4.2 Hz, H-5), 6.85 (1H, d, J=4.2 Hz, H-6)] showed downfield shifts of the H-3; H-3,4 and H-3,5 of 6,7 and 8 from shikimic acids, respectively, indicating that hydroxyl groups on the corresponding positions were esterified. Therefore, they were determined as 3-O-galloylshikimic acid (6), 3,4-di-O-galloylshikimic acid (7) and 3,5-di-O-galloylshikimic acid  $(8)^{(9)}$ , respectively.

Compounds 9-12 and 15-17 were identical with 1-O-galloyl- $\beta$ -D-glucose (9), corilagin (10), furosin (11), geraniin (12), gallic acid 3-O- $\beta$ -D-(6'-O-galloyl)-glucoside (15), kaempferol (16) and quercetin (17) by direct comparison with authen-

tic samples that were isolated from *Rosa taiwa-nensis*<sup>(10)</sup>, *Euphoria longana*<sup>(8)</sup>, *Macaranga sinensis*<sup>(11)</sup> or *M. tanarius*<sup>(12, 13)</sup> in our laboratory.

Compound 13 was obtained as a white amorphous powder, [α] D-91.6°. H-NMR spectrum showed signals consistent with a combination of an acetonyl group and a geraniin (12) possessing a five member ring DHHDP group. FABMS [m/z 991 (M-H)-] and physical data were agreeable with acetonylgeraniin A (13)<sup>(14)</sup>, and therefore the structure was identified.

The  ${}^{1}\text{H-NMR}$  spectrum pattern of 14 [ $\delta$  3.77  $(3H, s, -OCH_3)$ , 4.81 (1H, d, J = 7.5 Hz, glc H-1), 7.22, 7.37 (each 1H, d, J = 1.8 Hz, galloyl H-2, 6)] was similar to that of 15 except that 14 showed a galloyl group less, but a methoxyl signal more than 15. The signal of H-6 of the glucose moiety appeared at  $\delta$  3.75 (1H, dd, J=12.5, 4.6 Hz) and 3.84 (1H, dd, J=12.5, 4.6 Hz) indicated the C-6-OH was not esterified. Furth-ermore, the <sup>13</sup>C-NMR spectrum of 14 showed the downfield shift of the carbonyl carbon signal ( $\delta$  167.7) analogous to that ( $\delta$  169.8) observed in 15. This phenomenon indicated that the methyl group was conjugated to the carboxylic acid of gallic acid. Compound 14 was thus established as methyl gallate 3-O-β-D-glucoside. Gallic acid 3-O-β-D-glucoside has been reported by Kash-iwada et al in 1986<sup>(15)</sup>. This methylate may be an artifact.

E. moschatum contained galloylshikimic acid, 3-O-,3,4-di-O-and 3,5-di-O-galloylshikimic acid that differed from G. thunbergii<sup>(16)</sup>, which contained many kinds of galloylquinic acid, such as 3-O-,4-O-,5-O-and 3,4-di-O-galloylquinic acid. On the other hand, similar to G. thunbergii<sup>(17)</sup>, E. moschatum possessed many tannins including the active principle, geraniin.

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### 麝香牻牛兒苗之單寧及相關化合物之研究

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#### 摘 要

牻牛兒苗科植物 Geranium屬及 Erodium屬之地上部做爲止瀉整腸劑,有效成分爲單寧類之 Geraniin。麝香牻牛兒苗 (E. moschatum)爲在台灣新發現之馴化種。

麝香牻牛兒苗之含水丙酮萃取物,利用逆相層析膠質,分離到十七種單寧及相關化合物,包括五種phenolcarboxylic acid及酯類: protocatechuic acid (1), gallic acid (2), methyl gallate (3), caffeic acid(4), brevifolincarboxylic acid (5); 四種 gallotannins:3-O-galloylshikimic acid (6), 3,4-di-O-galloylshikimic acid (7), 3,5-di-O-galloylshikimic acid (8), 1-O-galloyl-b-D-glucose (9); 六種 ellagitannins及相關化合物:corilagin (10), furosin (11), geraniin (12), acetonylgeraniin A(13), methyl gallate 3-O-b-D-glucoside (14), gallic acid 3-O-b-D-(6'-O-galloyl)-glucoside (15)及二種flavonoids: kaempferol (16), quercetin (17)。化合物結構係依其物理性質及核磁共振光譜而決定,並與文獻或標準品比對確認。

關鍵詞:麝香牻牛兒苗、牻牛兒苗科、加水分解型單寧、類黃酮素。